

Antioxidant defense and secondary metabolites concentration in hyssop (*Hyssopus officinalis* L.) plants as affected by salt stress

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Abstract: Salt stress is one of the major limiting factors for plant production, and the quality of medicinal plants is also affected by soil salinity. Hyssop (*Hyssopus officinalis* L.) plants were cultivated for four weeks in perlite: sand and irrigated with Hoagland nutrient solution containing 0, 50, 100, 150, and 200 mM NaCl. Plants growth was decreased by salt stress while the leaf relative water content was not affected, and the chlorophyll content decreased only by the highest salt concentration (200 mM). Sodium was accumulated at small amounts, indicating a high ability of this species to exclude salt. Soluble sugars and proline were accumulated up to 1.6 and 4.5 fold, respectively. The antioxidant enzymes activity (peroxidase, catalase, ascorbate peroxidase) were increased by the salt treatments, particularly in the leaves. The levels of secondary metabolites (saponins, phenolics, flavonoids, anthocyanins, and iridoids) were all increased under salt stress, and the total antioxidant capacity of alcoholic extract of the leaves and roots was significantly higher in the salt-treated compared with control plants. Our results showed that hyssop is a salt-tolerant species, and the quality of this medicinal plant is improved when grown under saline conditions.

Key words: salinity; hyssop; *Hyssopus officinalis*; secondary metabolites; antioxidant enzymes

Antioksidativna obramba in vsebnost sekundarnih metabolitov v navadnem ožepku (*Hyssopus officinalis* L.) v razmerah solnega stresa

Izvleček: Solni stres je eden izmed dejavnikov, ki najbolj omejuje rast rastlin, v razmerah zasoljenih tal je prizadeta tudi kakovost zdravilnih rastlin. Navadni ožepok (*Hyssopus officinalis* L.) je bil gojen v mešanici perlita in peska in zalivan s Hoaglandovo hranilno raztopino, ki je vsebovala 0, 50, 100, 150, and 200 mM NaCl. Rast rastlin se je s solnim stresom zmanjšala, a relativna vsebnost vode v listih ni bila prizadeta in vsebnost klorofila se je zmanjšala le pri največji koncentraciji (200 mM NaCl). Natrij se je v rastlinah kopičil v majhnih količinah, kar nakazuje sposobnost te vrste, da izloča sol. Vsebnost topnih sladkorjev in prolina se je povečala za 1,6, oziroma 4,5 krat. Aktivnost antioksidacijskih encimov (peroksidaze, katalaze, askorbat peroksidaze) se je povečala po obravnavanjih s soljo, še posebej v listih. V razmerah solnega stresa se je povečala raven sekundarnih metabolitov (saponinov, fenolov, flavonoidov, antocianinov in iridoidov), celokupna antioksidacijska sposobnost alkoholnega ekstrakta listov in korenin je bila značilno večja pri rastlinah izpostavljenih soli kot pri kontroli. Rezultati so pokazali, da je navadni ožepok na sol strpna rastlina in, da se kakovost te zdravilne rastline izboljša, če jo gojimo v razmerah slanosti.

Ključne besede: slanost; navadni ožepok; *Hyssopus officinalis*; sekundarni metaboliti; antioksidacijski encimi

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1 INTRODUCTION

Soil salinity is one of the main abiotic stress factors threatening agricultural production worldwide. Salt in the soil is also considered the main factor limiting the dispersal of plants in their natural habitats (Acosta-Motos et al., 2017; Mushtaq et al., 2020). Salinity causes both osmotic and ionic stresses and affects all the plant major processes such as germination, photosynthesis, growth, water and nutrients balances and yield (Parida & Das, 2005; Parihar et al., 2015).

Salt stress affects the plant's primary metabolism and alters primary metabolites' concentrations, including soluble sugars and amino acids (Gupta and Huang, 2014). Among amino acids, proline plays a pivotal role in the plant's adaptation to salt stress by protecting cells from damages caused by excess accumulation of ions and salt-induced dehydration (Verbruggen and Hermans, 2008). As the second important group of compatible solutes, soluble sugars protect from dehydration and help sustain the structural integrity of plant cells under salt stress (Rosa et al., 2009).

Under environmental stress conditions such as salinity, the generation of higher reactive oxygen species (ROS) causes oxidative stress. It results in membrane damage characterized by elevated levels of malondialdehyde (MDA). Plants employ defensive systems for scavenging ROS and protecting from damaging oxidative reactions through different antioxidant enzymes such as peroxidases (POD), catalase (CAT), and ascorbate peroxidase (APX) (Foyer et al., 1994; Gupta and Huang, 2014; Akyol et al., 2020). Proline protects plant cells from ionic and osmotic stresses and contributes to the scavenging ROS such as hydroxyl radicals (Verbruggen and Hermans, 2008).

In addition to primary metabolism, secondary plant metabolism is also influenced by salt stress (Ahmad and Sharma, 2008). The concentration of secondary metabolites highly depends on plants' growth stage, especially environmental conditions, including light intensity and stress factors such as salt (Ahl and Omer, 2011). Salt stress has a positive or negative impact on secondary metabolites' biosynthesis depending on plant species or the severity of stress (Verma and Shukla, 2015). Salt stress led to about 8–35 % increase in total phenolics and about 35 % increase in total flavonoid content in *Portulaca oleracea* L. (Alam et al., 2015). An enhancement of antioxidant activity and flavonoid and phenolics contents has

also been observed in *Cichorium spinosum* L. under salt stress (Petropoulos et al., 2017).

In the members of Lamiaceae, salinity may cause substantial changes in the compositions and yield of secondary metabolites (Taarit et al., 2009). Salt stress led to about 20–40 % increase in the total phenolics and flavonoids content in *Thymus* species (Zrig et al., 2016). Salt stress significantly induced the biosynthesis of some crucial essential oil and phenolic compounds in *Salvia mirzayanii* Rech.f. & Esfand. (Valifard et al., 2014). Considering the medicinal application of most plant species from Lamiaceae, it is important to know the effect of salt stress on the quality and quantity of secondary metabolites. The content of secondary metabolites is also a determining factor for the total antioxidant activity of plant extract defined either by FRAP (Ferric Reducing Ability of Plasma) or through DPPH (2, 2-diphenyl-1-picrylhydrazyl) assay (Fukumoto and Mazza, 2000; Sethi et al., 2020).

Hyssop (*Hyssopus officinalis* L.) is a perennial subshrub belonging to the Lamiaceae family and distributes in the eastern Mediterranean to central Asia (Fathiazad and Hamedeyazdan, 2011). Hyssop is a popular medicinal herb with carminative, tonic, antiseptic, and expectorant properties and is used to remedy congestion, lung complaints, and cardiovascular disorders (Hristova et al., 2015). Extracted essential oils from the shoots of hyssop plants possess a unique aroma and are widely used in the food, pharmaceutical, and cosmetics industries (Kazazi et al., 2007). Different polyphenolic compounds identified in this species contain different glycones and aglycones such as flavonoids, quercetin, apigenin, diosmin, luteolin, and other phenolic compounds such as chlorogenic, ferulic, protocatechuic, syringic, caffeic, and p-hydroxybenzoic acids (Fathiazad and Hamedeyazdan, 2011).

Hyssop is a xerophyte species and is well adapted to drought conditions (Khazaie et al., 2008). Effect of salt stress on the activity of antioxidant enzymes has been investigated in this species (Jahantigh et al., 2016). There is no report, however, on the secondary metabolites levels under salt stress in hyssop plants. This work aimed to investigate the effect of salinity on growth, ROS scavenging activity, the content of various secondary metabolites, and the antioxidant capacity of the leaf and root extract in this species.

2 MATERIAL AND METHODS

2.1 PLANT MATERIAL AND TREATMENTS

Seeds of hyssop (*H. officinalis*) were purchased from

Pakan-Bazr Company (Isfahan, Iran). The seeds were surface-sterilized using 1 % sodium hypochlorite then were sown in the 2 L pots (15 seeds in each pot) containing sterilized perlite: sand (1:3) mixture and placed at 4 °C for stratification. After 4 days, the pots were transferred to the greenhouse conditions at 25/22 °C day/night temperature regimes, at a 16/8 h day/night cycle, and relative humidity of 60 %. After germination, plants were irrigated with 20 % Hoagland solution once a week and with 100 ml distilled water twice a week throughout the experiment.

Salt stress was imposed at a four-leaf stage with four NaCl concentrations (50, 100, 150, and 200 mM) applied with irrigation water to the pots gradually within one week. Plants were grown for four weeks after starting salt stress and then were harvested. At harvest, fresh mass (FM) of shoot and roots were determined, and subsamples were taken and immediately frozen in liquid nitrogen and then stored at -80 °C until analysis. Another group of samples was oven-dried and, after determination of dry mass (DM), were used for the analysis of secondary metabolites and elemental concentrations.

2.2 BIOCHEMICAL MEASUREMENTS

Soluble carbohydrate content was determined according to the phenol-sulfuric acid method (Dubois et al., 1956). Proline was quantified according to the methods of Bates et al. (1973), and the content of soluble proteins was determined using the Bradford method (Bradford, 1976) with bovine serum albumin as standard.

To determine the leaf content of chlorophyll (Chl) and carotenoids, 0.2 g of fresh leaf samples were homogenized with 2 ml of 80 % acetone and centrifuged at 4000 g for 10 min. After that, the samples' absorbance was recorded by a spectrophotometer (Bausch & Lomb 70) at 663, 645, and 480 nm, and the contents of pigments were calculated according to the following equations where A corresponds to the absorbance (Flores-de-Santiago et al., 2016):

$$\text{Chl a} = 12.21A_{663} - 2.81A_{646}$$

$$\text{Chl b} = 20.13A_{646} - 5.03A_{663}$$

$$\text{Carotenoids} = [1000A_{470} - 3.27(\text{Chl a}) - 104(\text{Chl b})] / 229$$

The leaf relative water content (RWC, %) was measured according to the following equation:

$$\text{RWC} (\%) = [(FM - DM) / (TM - DM)] \times 100$$

For determination of turgid mass (TM), leaf disks (5 mm diameter) were submerged for 5 h in distilled water, thereafter, they were blotted dry gently on a paper towel and weighed.

2.3 OXIDATIVE STRESS MARKERS AND ACTIVITIES OF ANTIOXIDANT ENZYMES

Sergiev et al. (1997) method with slight modification was used for quantification of H₂O₂ content. 100 mg of samples were ground in liquid nitrogen, extracted with trichloroacetic acid (TCA) in an ice bath, centrifuged at 13,000 g for 15 min. The 500 µl of supernatant was added to potassium phosphate buffer (pH 7.0), and the H₂O₂ content was determined based on supernatant absorbance at 390 nm. Malondialdehyde (MDA) content was evaluated by the method of Heath and Packer (1968). 100 mg of samples were homogenized in 1 ml 0.1 % (v/v) TCA and centrifuged at 12,000 g for 10 min. The extracted supernatant (0.1 ml) was mixed with 20 % TCA containing 0.5 % (w/v) thiobarbituric acid (TBA). The mixture was incubated at 100 °C for 10 min then centrifuged at 10,000 g for 15 min. TBA reactive substances' content was calculated based on the difference in absorbance at 532 and 600 nm using an extinction coefficient of 155 mM⁻¹ cm⁻¹.

For assay of catalase (CAT), peroxidase (POD), and ascorbate peroxidase (APX), 100 mg of the samples was extracted in 5 ml of 100 mM phosphate buffer. The homogenate was then centrifuged for 10 min and used for enzyme assays. CAT activity was analyzed by Beers and Sizer (1952) method. The 2 ml of reaction mixture containing 100 mM potassium phosphate buffer (pH 7.0) was mixed with 400 µl of 6 % H₂O₂ and 100 µl of enzyme extract. CAT activity was calculated to reduce the H₂O₂ absorption at a wavelength of 240 nm using an extinction coefficient of 0.036 mM⁻¹ cm⁻¹. POD activity was measured by the method of Lin and Kao (1999). The reaction mixture (2 ml) contained potassium phosphate buffer (50 mM, pH 7), guaiacol solution (9 mM), H₂O₂ (19 mM) and root or leaf extract (100 µl). POD activity calculated the absorbance changes at 470 nm using the extinction coefficient of 26.6 mM⁻¹ cm⁻¹. For assay of APX, the reaction mixture contained 250 mM phosphate buffer (pH 7), 1.2 mM H₂O₂, 0.5 mM ascorbic acid, and 0.1 mM EDTA. The reaction was started by adding H₂O₂ to the mixture. Total APX activity was calculated to reduce the absorbance at 290 nm for 2 min using the extinction coefficient of ascorbic acid (2.8 mM⁻¹ cm⁻¹) (Dazy et al., 2008).

2.4 SECONDARY METABOLITES QUANTIFICATION

For analyzing secondary metabolites, 100 mg of leaf or root dried samples were dissolved in 80 % ethanol (5 ml) and sonicated for 20 min at room temperature. The resulted mixture was centrifuged at 3,000 g for 15 min.

The extraction was repeated three times, and the supernatants were pooled and stored until analysis.

A colorimetric method was used to determine the total amount of saponins (Hiai et al., 1975). In a test tube, 0.5 ml of plant ethanol extract was mixed with 0.5 ml of vanillin and 5 ml of 72 % sulfuric acid. The mixture was shaken and heated for 10 min at 60 °C in a water bath. After cooling in the water at room temperature, the extract's absorbance was determined spectrophotometrically at 545 nm. To determine the concentration of the total phenolics, 2.5 ml of Folin-Ciocalteu-Deniz indicator and 2.5 ml of 2 % sodium carbonate solution were added to 0.5 ml plant ethanol extract. The resulting mixture was homogenized and incubated in the dark for 30 min. The absorbance of the solution was measured spectrophotometrically at 750 nm (Seevers et al., 1971). The total flavonoids concentration was determined using the aluminum chloride colorimetric method (Zhishen et al., 1999). The extract's 0.5 ml was mixed with 4.5 ml of distilled water and 0.5 ml of 5 % sodium nitrite solution. After 5 min, 0.5 ml of 10 % aluminum chloride was added, and the mixture was incubated for 6 min. After that, 4 ml of 1 M NaOH was added, and after 15 min, the absorbance of the mixture was read at 510 nm by a spectrophotometer. Quercetin was used for the creation of a standard curve. For analyzing the total anthocyanins concentration, 0.2 ml of alcohol extract was diluted separately with 4.8 ml potassium chloride (pH 1) and sodium acetate buffer (pH 4.5). The solutions were incubated in the dark for 15 min. The absorbance of both groups of samples was determined at 510 and 700 nm, and the concentration of anthocyanins (A) was calculated according to the following equation and reported as cyanidin-3-O-glucoside equivalent using an extinction coefficient of 26.9 mM⁻¹ cm⁻¹ (Giusti and Wrolstad, 2001):

$$A = [(A_{510 \text{ nm}} - A_{700 \text{ nm}})_{\text{pH } 1.0} - [(A_{510 \text{ nm}} - A_{700 \text{ nm}})_{\text{pH } 4.5}]$$

A colorimetric method based on the color reaction of cobin with glycine was used to determine the total iridoid concentration (Narayanan and Akamanchi, 2003). One ml of ethanol extract was mixed with 2 ml distilled water, 1 ml of 10 % glycine, and 1 ml sulfuric acid (0.1 M). The mixture was shaken and heated in a water bath (95 °C) for 1 h. After cooling at room temperature, the absorbance was measured at 554 nm.

2.5 DETERMINATION OF TOTAL ANTIOXIDANT ACTIVITY

Different plant extract concentrations in methanol (10, 25, 50, 100, 250, and 500 µg ml⁻¹) were prepared and incubated with freshly-prepared 80 µg ml⁻¹ DPPH (2,

2-diphenyl-1-picrylhydrazyl). The mixtures were shaken and placed in the dark for 30 min, then the absorbance of the samples in parallel with a solution without plant extract (as blank) was read at 517 nm. The inhibition (%) of DPPH radical formation was calculated according to the following equation:

$$\text{Inhibition (\%)} = [(A_{\text{blank}} - A_{\text{sample}}) / A_{\text{blank}}] \times 100$$

Plant extract's antioxidant activity was reported as the sample concentration providing 50 % inhibition (IC₅₀) calculated by plotting inhibition percentages against the samples' concentration (Sarkar et al., 2006).

2.6 DETERMINATION OF ELEMENTS CONCENTRATIONS

To determine potassium (K) and sodium (Na) concentrations, 100 mg milled oven-dried samples were digested in concentrated nitric acid overnight, then heated at 80 °C for one hour and dissolved in 1 % HCl. The concentrations of K and Na were determined by flame photometry (Kalra, 1997).

2.7 STATISTICAL ANALYSIS

The experiment was undertaken as a completely randomized block design with three pots as independent replicates for each treatment. Data were presented as means ± standard deviation (SD). Comparison of means was performed by Tukey test ($p < 0.05$) using SPSS (version 23, for Windows; SPSS Inc., Chicago, IL, USA).

3 RESULTS AND DISCUSSION

Plants' growth was expectedly decreased under salt stress conditions (Fig. 1). A significant effect of salt stress on the shoot and root fresh weight was observed at 100 mM salt and higher. Dry biomass of plants was depressed up to 75 % at a salt concentration of 200 mM. Shoot height and length of the taproot significantly decreased by 50 mM salt and higher (Fig. 2).

The leaf content of Chl a and Chl b were not affected by salt concentration up to 150 and 100 mM NaCl, respectively. Leaf carotenoid content and RWC, were not significantly influenced by applied salt levels (Table 1).

Activities of all three analyzed antioxidant enzymes were higher in salt-stressed plants both in the leaves and roots (Fig. 3). At 200 mM salt concentration, the leaf activities of POD, CAT and APX increased up to 2.3, 6.3 and 6.9 fold, respectively, compared to the control treat-



Figure 1: Hyssop (*Hyssopus officinalis* L.) plants grown for 4 weeks under control or different salt concentrations (50, 100, 150 and 200 mM NaCl) under greenhouse conditions.

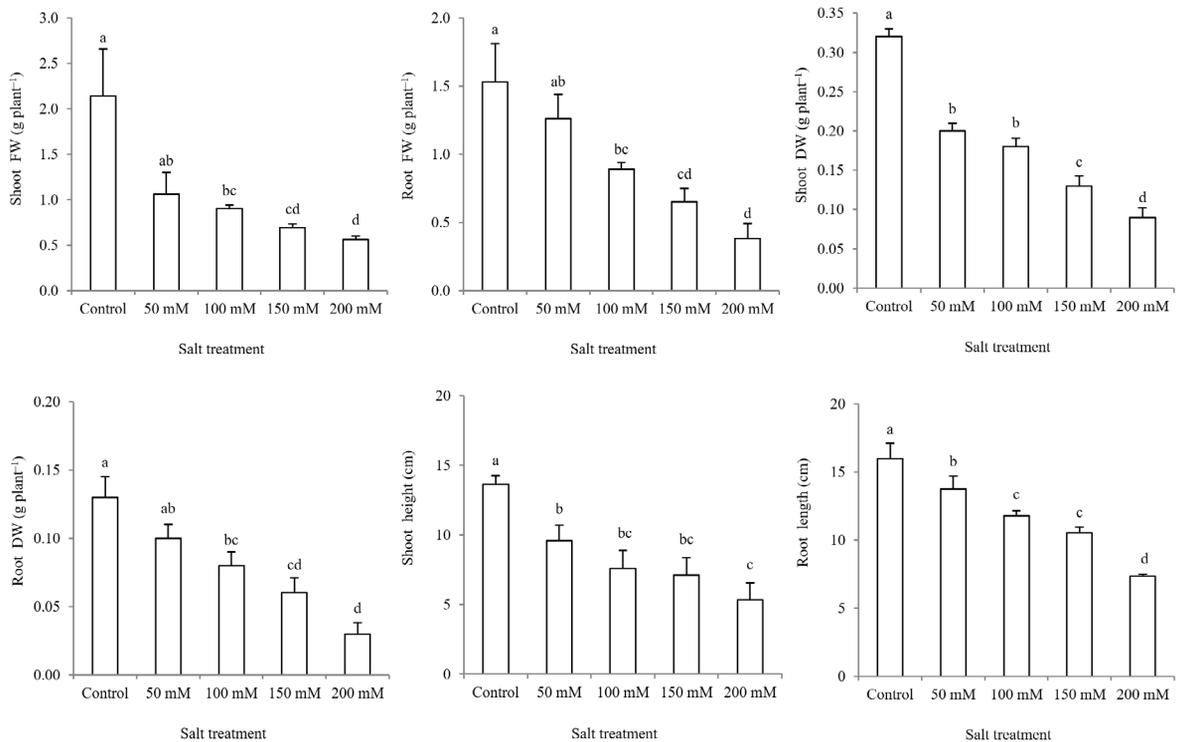


Figure 2: Fresh and dry biomass (g plant⁻¹), shoot height and root length (cm) of hyssop (*Hyssopus officinalis* L.) plants grown for 4 weeks under control or different salt concentrations (50, 100, 150 and 200 mM NaCl) under greenhouse conditions. Bars indicated by the same letter are not significantly different ($p < 0.05$).

Table 1: Content of chlorophyll (Chl) *a*, *b* and carotenoids (mg g⁻¹ FM) and relative water content (RWC, %) in the leaves of hyssop (*Hyssopus officinalis* L.) plants grown for 4 weeks under control or different salt concentrations (50, 100, 150 and 200 mM NaCl) under greenhouse conditions. Data of each column indicated by the same letter are not significantly different ($p < 0.05$).

NaCl concentration	Chl <i>a</i>	Chl <i>b</i>	Carotenoids	RWC
Control	3.01 ± 0.27 ^a	1.27 ± 0.18 ^a	4.95 ± 0.53 ^a	0.58 ± 0.05 ^a
50 mM	2.95 ± 0.21 ^a	1.07 ± 0.06 ^a	5.40 ± 3.74 ^a	0.50 ± 0.11 ^a
100 mM	2.18 ± 0.04 ^{ab}	1.09 ± 0.01 ^a	3.74 ± 0.13 ^a	0.55 ± 0.12 ^a
150 mM	2.48 ± 0.19 ^a	0.74 ± 0.07 ^b	4.19 ± 0.35 ^a	0.46 ± 0.04 ^a
200 mM	1.36 ± 0.79 ^b	0.76 ± 0.09 ^b	2.51 ± 0.25 ^a	0.49 ± 0.03 ^a

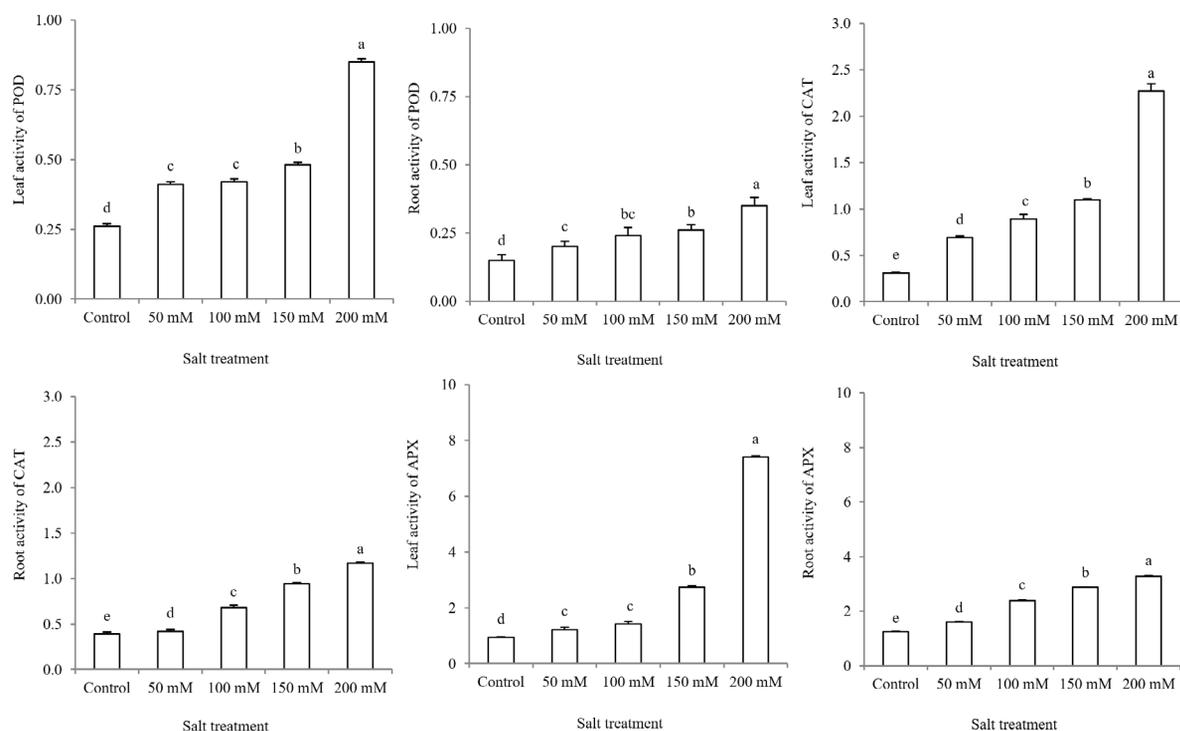


Figure 3: Activity of peroxidase ($\mu\text{mol mg}^{-1} \text{protein min}^{-1}$), catalase ($\mu\text{mol mg}^{-1} \text{protein min}^{-1}$) and ascorbate peroxidase ($\mu\text{mol mg}^{-1} \text{protein min}^{-1}$) in the leaves and roots of hyssop (*Hyssopus officinalis* L.) plants grown for 4 weeks under control or different salt concentrations (50, 100, 150 and 200 mM NaCl) under greenhouse conditions. Bars indicated by the same letter are not significantly different ($p < 0.05$).

ment. In the roots, the activities of POD, CAT and APX were 1.3, 2.0 and 1.6 fold higher than the control plants (Fig. 3).

The soluble sugar content increased gradually in response to increasing salt levels in the medium, both in the leaves and roots. The extent of the increase was higher for the leaves (60 % at 200 mM salt) than the roots (20 % at 200 mM salt). The content of soluble proteins decreased by salt stress both in the leaves and roots. However, different salt levels did not differ in their effect on the root protein content, while in the leaves, it was continuously decreased by increasing salt levels (Table 2). Proline was accumulated both in the leaves and roots upon exposure

to salt stress. In the leaves, proline content responded to low salt level (50 mM) and accumulated up to 4.5 fold in the presence of 200 mM salt. In comparison, in the roots, salt's significant effect was not observed at a low level (50 mM) and accumulated to much less extent, i.e., 1.8 fold under 200 mM salt. The concentration of K was steadily decreased under salt treatment while that of Na increased both in the leaves and roots (Table 2).

Leaf content of MDA was increased by salt stress both in the leaves and roots. The treatment effect was more prominent in the roots with up to 2.3 fold MDA accumulation at the salt treatment of 200 mM, while the corresponding value for the leaves was only 1.2 fold. The

Table 2: Contents of soluble sugars (mg g^{-1} FM), soluble proteins (mg g^{-1} FM), proline ($\mu\text{mol g}^{-1}$ M), potassium and sodium (mg g^{-1} DM) in the leaves and roots of hyssop (*Hyssopus officinalis* L.) plants grown for 4 weeks under control or different salt concentrations (50, 100, 150 and 200 mM NaCl) under greenhouse conditions. Data of each column indicated by the same letter are not significantly different ($p < 0.05$).

NaCl concentration	Soluble sugars	Soluble proteins	Proline	Potassium	Sodium
Control	21.43 ± 0.84 ^d	12.3 ± 0.02 ^a	5.03 ± 0.01 ^e	4.72 ± 0.15 ^a	2.39 ± 0.09 ^d
50 mM	23.05 ± 1.27 ^{cd}	10.7 ± 0.10 ^b	5.88 ± 0.01 ^d	4.39 ± 0.06 ^b	3.63 ± 0.24 ^c
100 mM	26.23 ± 1.67 ^{bc}	9.61 ± 0.23 ^c	6.53 ± 0.05 ^c	4.26 ± 0.01 ^b	4.13 ± 0.13 ^c
150 mM	28.27 ± 0.32 ^b	9.04 ± 0.36 ^d	12.2 ± 0.10 ^b	3.66 ± 0.01 ^c	4.84 ± 0.26 ^b
200 mM	34.05 ± 1.90 ^a	7.01 ± 0.07 ^e	22.5 ± 0.03 ^a	3.29 ± 0.15 ^d	5.58 ± 0.40 ^a
			Roots		
Control	31.27 ± 0.16 ^b	10.8 ± 1.64 ^a	2.70 ± 0.27 ^d	7.16 ± 0.10 ^a	0.35 ± 0.01 ^e
50 mM	31.41 ± 0.07 ^b	8.44 ± 0.02 ^b	2.94 ± 0.20 ^{cd}	6.56 ± 0.10 ^b	0.59 ± 0.01 ^d
100 mM	33.51 ± 0.12 ^b	7.51 ± 0.02 ^b	3.57 ± 0.13 ^{bc}	6.26 ± 0.10 ^c	0.75 ± 0.07 ^c
150 mM	36.63 ± 1.81 ^a	7.14 ± 0.05 ^b	4.05 ± 0.14 ^b	5.46 ± 0.10 ^d	1.05 ± 0.02 ^b
200 mM	37.51 ± 0.75 ^a	6.76 ± 0.20 ^b	4.79 ± 0.37 ^a	4.36 ± 0.10 ^e	1.42 ± 0.07 ^a

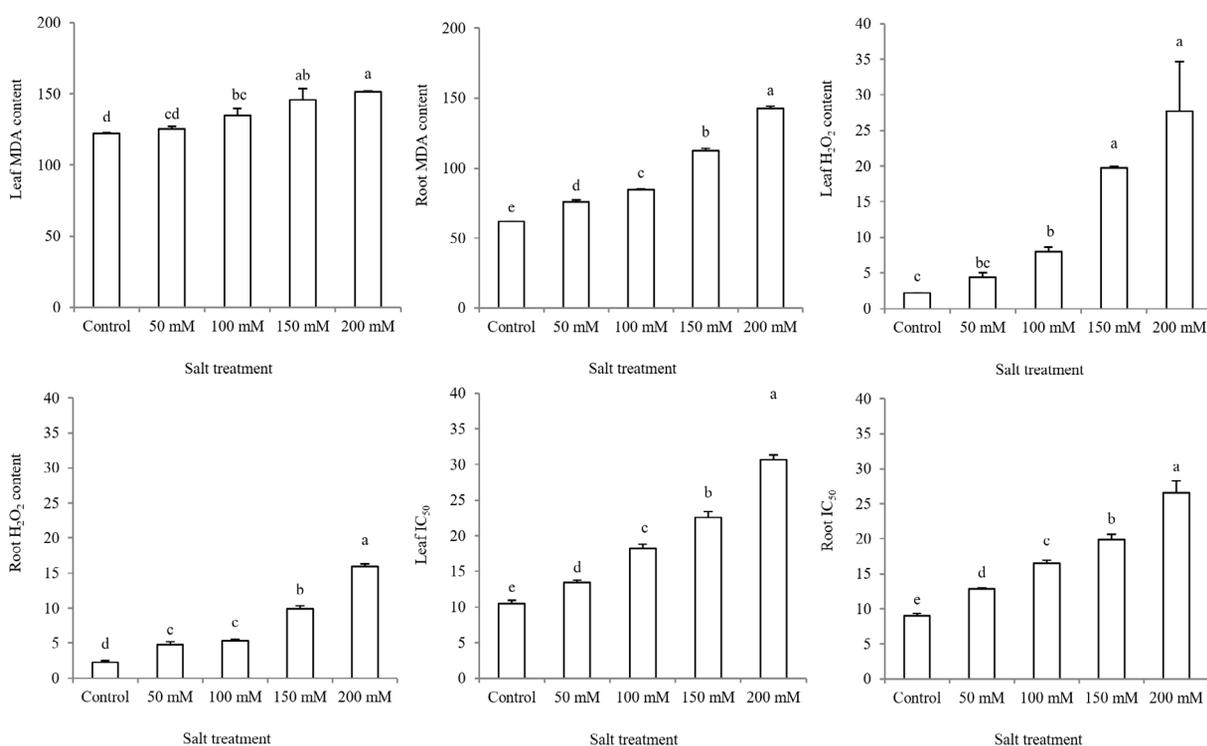


Figure 4: Contents of malondialdehyde (MDA, $\mu\text{mol g}^{-1}$ FM), H_2O_2 ($\mu\text{mol g}^{-1}$ FM) and total antioxidant activity (IC_{50}) in the leaves and roots of hyssop (*Hyssopus officinalis* L.) plants grown for 4 weeks under control or different salt concentrations (50, 100, 150 and 200 mM NaCl) under greenhouse conditions. Bars indicated by the same letter are not significantly different ($p < 0.05$).

Table 3: Concentrations of saponins (mg g⁻¹ DM), total phenolics (mg g⁻¹ DM), total flavonoids (µg g⁻¹ DM), anthocyanins (µg g⁻¹ DM) and iridoids (mg g⁻¹ DM) in the leaves and roots of hyssop (*Hyssopus officinalis* L.) plants grown for 4 weeks under control or different salt concentrations (50, 100, 150 and 200 mM NaCl) under greenhouse conditions. Data of each column indicated by the same letter are not significantly different ($p < 0.05$).

NaCl concentration	Saponins	Phenolics	Flavonoids	Anthocyanins	Iridoids
	Leaves				
Control	330 ± 2.8 ^c	19.3 ± 0.10 ^d	75.4 ± 3.96 ^d	10.15 ± 0.57 ^d	50.7 ± 1.59 ^d
50 mM	371 ± 24.7 ^c	22.1 ± 0.98 ^c	87.6 ± 3.22 ^c	18.70 ± 0.62 ^c	55.0 ± 2.57 ^{cd}
100 mM	475 ± 83.6 ^c	24.4 ± 0.33 ^b	93.9 ± 1.83 ^c	20.55 ± 0.78 ^c	73.7 ± 3.13 ^{bc}
150 mM	644 ± 54.6 ^b	25.7 ± 0.90 ^{ab}	109 ± 4.58 ^b	26.25 ± 0.83 ^b	92.2 ± 5.78 ^b
200 mM	801 ± 68.9 ^a	27.4 ± 0.48 ^a	118 ± 2.37 ^a	50.09 ± 1.98 ^a	129.1 ± 15.2 ^a
	Roots				
Control	390 ± 23.7 ^c	45.31 ± 1.22 ^d	199 ± 18.31 ^c	0.90 ± 0.10 ^d	70.5 ± 0.33 ^a
50 mM	446 ± 10.1 ^c	48.79 ± 0.82 ^{cd}	224 ± 1.64 ^{bc}	1.50 ± 0.10 ^d	75.3 ± 0.98 ^a
100 mM	610 ± 109 ^b	52.75 ± 0.38 ^{bc}	239 ± 8.38 ^b	2.52 ± 0.50 ^c	76.8 ± 0.18 ^a
150 mM	755 ± 18.3 ^a	55.03 ± 0.43 ^b	249 ± 1.77 ^{ab}	4.00 ± 0.56 ^b	78.5 ± 0.64 ^a
200 mM	859 ± 5.81 ^a	59.39 ± 3.10 ^a	267 ± 9.39 ^a	6.98 ± 0.07 ^a	75.9 ± 6.79 ^a

content of H₂O₂ was consistently increased by increasing salt level. The H₂O₂ accumulation in response to higher salt levels (150 and 200 mM) was more prominent in the leaves than in the roots. The total antioxidant activity (IC₅₀) was increased up to 2.9 fold in the salt-stressed plants (Fig. 4).

The leaf and root concentrations of all analyzed secondary metabolites were increased by exposure to salt stress in the leaves and roots except iridoids in the roots that remained unchanged (Table 3). Lower salt level (50 mM) was effective in the increasing phenolics, flavonoids, and anthocyanins in the leaves, while in the roots, a significant effect was observed by higher salt level (100 mM). The extent of salt-induced increase in the concentration of analyzed secondary metabolites was in the range of 1.5-2.5 fold except for anthocyanins. This metabolite showed up to 4.9 and 7.8 fold increase upon exposure to 200 mM salt in the leaves and roots, respectively (Table 3).

4 DISCUSSION

4.1 EFFECT OF SALT STRESS ON GROWTH, NA CONCENTRATION AND LEAF CHL CONTENT

Hyssop is a drought-tolerant species (Khazaie et al., 2008); however, its salt tolerance has not been studied so far. Our data demonstrated that hyssop is also tolerant to salinity stress as the plants in our study survived after 4 weeks of salt treatment of 200 mM. Such high salt toler-

ance has been rarely reported in the members of Lamiaceae. In the studies on the salt tolerance in other Lamiaceae species such as *Thymus*, *Perilla*, and *Salvia*, much higher growth inhibition by salt has been reported, and plants were killed by salt concentrations higher than 100 mM (Paiva et al., 2018; Bistgani et al., 2019; Salachna et al., 2019).

The Na concentration data showed that this species is a Na-excluder salt-tolerant plant and can avoid root Na uptake. The low Na accumulation in the roots and leaves was accompanied by stable amounts of RWC showing that this species maintains tissue water content despite exposure to low water potentials in the rooting medium. On the other hand, a constitutively lower RWC (0.49-0.58 %) shows that these species cope with low water potentials through passive water content reduction. A similar mechanism for salt tolerance has been observed in *Thellungiella*, a halophyte close relative of *Arabidopsis* (Lugan et al., 2010).

In agreement with the conclusion mentioned above on high salt tolerance in hyssop plants, leaf Chl content remained unaffected by salt treatment up to 150 mM suggesting that leaf photosynthetic capacity remained mainly unaffected under these conditions. The maintenance of photosynthesis and carbon metabolism may help plants retain an ability to synthesize organic osmolytes, including soluble sugars and proline (Chaves et al., 2009). These two organic osmolytes were accumulated in the leaves up to 1.6 and 4.5 fold, respectively, which may contribute significantly to plants' osmotic homeostasis under salt stress conditions. In addition to osmotic functions,

these osmolytes contribute to protecting cell structures, ROS scavenging, and nitrogen and carbon sources under stress conditions (Verbruggen and Hermans, 2008; Mattioli et al., 2009; Rosa et al., 2009).

4.2 EFFECT OF SALT STRESS ON THE ACTIVITY OF ROS ACCUMULATION, SCAVENGING AND MEMBRANE INTEGRITY

The activities of ROS scavenging enzymes were expectedly increased by salt treatment both in the leaves and roots. The salt-induced activity of all three analyzed enzymes was higher in the leaves compared with the roots that may contribute to high protection of leaves against salt-induced damage. Better protection of leaves than roots was confirmed by the maintenance of a high Chl content under high salinity treatments and much less increase of MDA content under salt stress (24 % at 200 mM salt) compared with the roots (130 % at 200 mM salt).

Nevertheless, the accumulation of H_2O_2 was higher in the leaf than in the roots indicating that higher enzyme activities were not sufficient for inhibition of H_2O_2 accumulation in the leaves. Although H_2O_2 belongs to ROS, it is known to be much less damaging in comparison to superoxide and hydroxyl radicals (Cheng et al., 2006). The prevailing effect of H_2O_2 is a signaling role. It has been observed that H_2O_2 is an important signal that is raised under salt stress and is responsible for the activation of various defense pathways in salt-stressed plants (Shu-Hsien et al., 2005). We suggest that the higher capability of leaves for H_2O_2 accumulation, and activation of defense pathways may be partly responsible for higher protection of leaves against salt stress than the roots.

4.3 EFFECT OF SALT STRESS ON THE CONCENTRATION OF SECONDARY METABOLITES

The concentration of all analyzed secondary metabolites was higher in the salt-stressed hyssop plants in our study. The effect of salt treatment on the levels of phenolics, flavonoids, and anthocyanins has been reported in other Lamiaceae species (Kotagiri et al., 2017; Bistgani et al., 2019; Salachna et al., 2019; Becerra-Gudiño et al., 2019). However, in hyssop, the quantity of bioactive compounds as affected by salinity has not been investigated so far. Our study is also the first report on the effect of salt stress on the saponins and iridoids. Iridoids are a type of monoterpenoids found in plants, mainly as glyco-

sides (Wang et al., 2020). The iridoids produced by plants act as a defense against herbivores or microorganisms (Fuchs et al., 2004). From a medicinal point of view, these compounds have wound-healing and anti-inflammatory effects with therapeutic potential for Alzheimer's and Parkinson's diseases (Dinda et al., 2019; Hussain et al., 2019). Saponins with one or more hydrophilic glycoside moieties combined with a lipophilic triterpene molecule (El Aziz et al., 2019) exhibit medicinal properties such as hemolytic factor, anti-inflammatory, antibacterial, anti-fungal, antiviral, anticancer, and cholesterol-lowering action in animals and human (Sparg et al., 2004). Besides, saponins formed the backbone of modern medicine or drugs and were considered a starting precursor for the semi-synthesis of steroidal drugs in the pharmaceutical industry (Netala et al., 2015).

It is noteworthy that higher concentrations of the secondary metabolites accompanied by reduction of biomass suggests a 'concentration-effect' in our hyssop plants. Nonetheless, it indicates that salt treatment did not inhibit the secondary metabolism in this species.

4.4 EFFECT OF SALT STRESS ON THE ANTIOXIDANT ACTIVITY OF LEAF AND ROOT EXTRACT

The DPPH scavenging activity is defined as the antioxidant activity of food and medicinal plants (Fukumoto and Mazza 2000; Sethi et al., 2020), has been reported for hyssop plants (Fathiazad et al., 2011; Pirbalouti et al., 2019; Rezaei Savadkouhi et al., 2020). However, the effect of salt on this parameter has not been studied so far. Here in our work, the DPPH free radical scavenging activity was increased by salt treatment for the leaf and root extracts. Electron donation is an important mechanism in which plants bioactive compounds convert free radicals to nonradical forms and thus, end the radical chain reactions (San Miguel-Chávez, 2017; Shahidi and Ambigaipalan, 2015). By analyzing various plant species, it has been observed that the main component of DPPH scavenging activity is phenolics, flavonoids, and anthocyanins (Fukumoto and Mazza 2000; Kim et al., 2007). Phenolic compounds act as a reducing agent and a hydrogen donor and show antioxidant effects (Oke et al., 2009).

5 CONCLUSION

Our data demonstrated that, hyssop plants are a salt-tolerant species, and secondary metabolites are increased upon growth under salinity. Regarding the fact

that, the plants dry matter production was reduced under higher salt levels. i.e., 200 mM equivalent with 14.5 dS m⁻¹, cultivation of this species is recommended in the soils with electrical conductivity up to 10 dS m⁻¹. Thus, the cultivation of this species on salinized soils that are unsuitable for most crop species is an alternative for low-income farmers.

6 REFERENCES

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